

# **Materials for High-Temperature Thermochemical Processes**

**D. F. Wilson & W. R. Corwin, Oak Ridge National Laboratory**

**S. Sherman & J. Kolts, Idaho National Laboratory**

**P. Pickard, Sandia National Laboratory**

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# NHI Materials for TC Processes Task Focuses on Planning & Coordinating

## Timeline

- ❖ Project start date FY04
- ❖ Project end date FY08
- ❖ Percent complete ~10%

## Budget

- ❖ Total project funding – 1M\$
  - ◆ DOE share: 1M
  - ◆ Funding received in FY04: 0.05M
- ❖ FY05 Funding: 0.1M

## Barriers

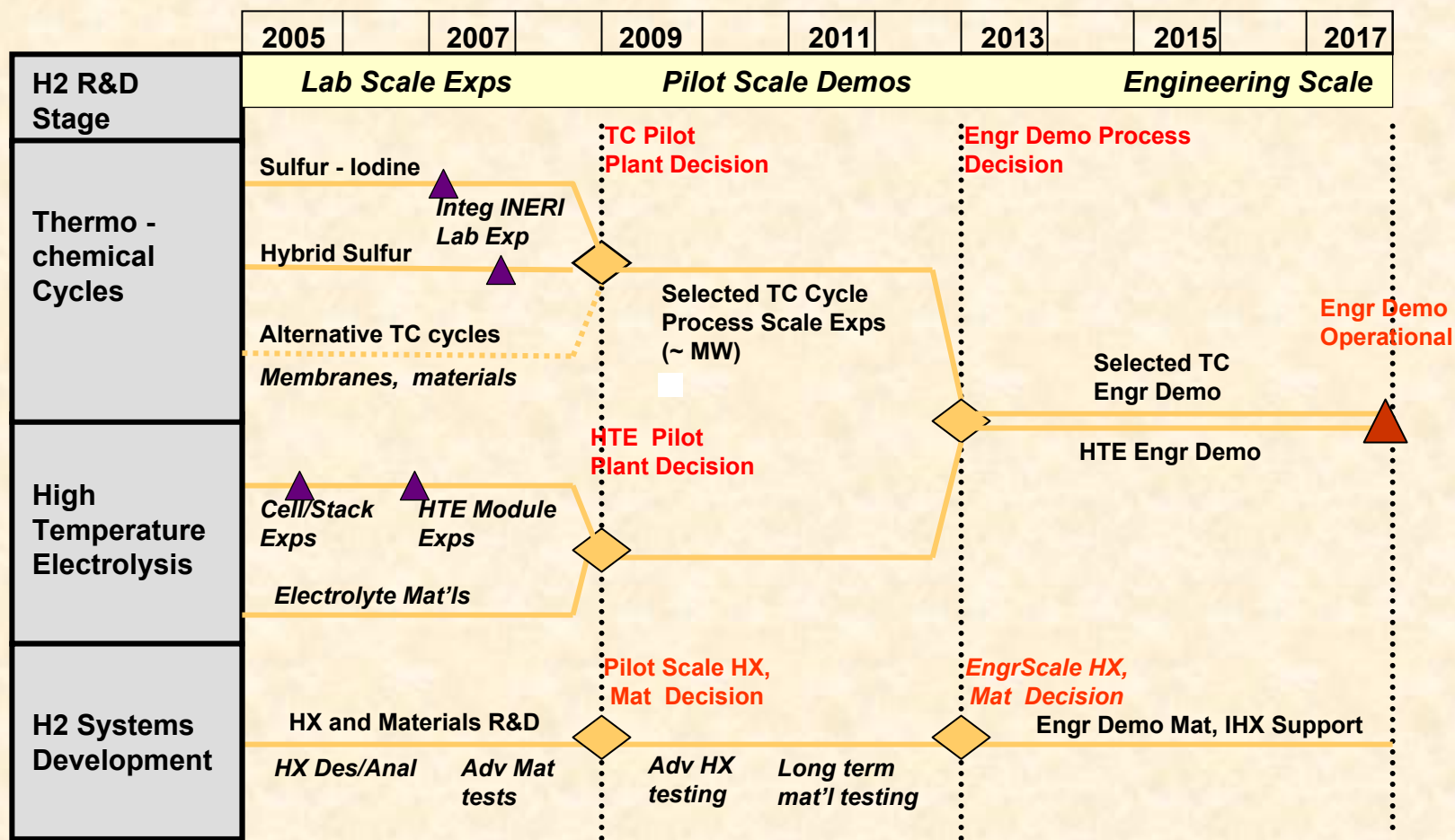
- ❖ Thermochemical (TC) process service environments
  - ◆ High temperature
  - ◆ Corrosive
  - ◆ Lack of sufficient testing capabilities for TC conditions

## Partners

### NHI TC Program

- ❖ Industry (GA, Cerametec)
- ❖ Universities (UNLV, MIT, UCB)
- ❖ Labs (INL, ORNL, SNL)

# NHI Materials for TC Processes Plans and Coordinates R&D through FY 2008



# Objectives of Materials for High-Temperature Thermochemical Processes Are:

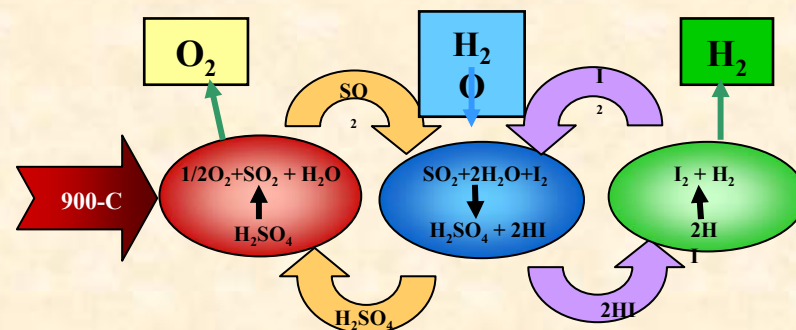
- ❖ Assess range of service conditions for NHI Thermochemical processes (Sulfur cycles, Ca-Br, others)
- ❖ Identify candidate materials of construction for cycle components (alloys, ceramics, refractories)
- ❖ Develop materials testing approach and priorities to support NHI TC cycle development (NHI Materials Development and Testing Plan)
- ❖ Coordinate materials planning for NHI and monitor “evolving” research and development activities

*System Interface and High Temperature Heat Exchanger Programs at INL & UNLV are part of this activity (S. Sherman, INL, coordinates)*



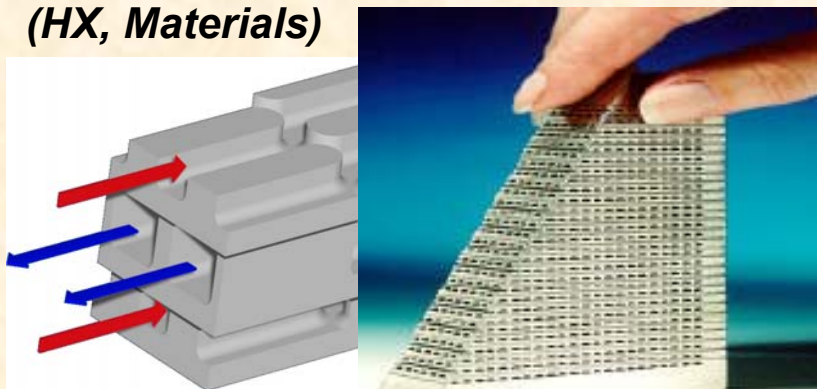
# Thermochemical Cycle R&D Areas Include:

- ❖ Thermochemical Cycles (*Scaling, efficiency*)
  - ◆ Sulfur cycles (*S-I, Hybrid*)
  - ◆ Alternate, *Ca-Br*
- ❖ System Interface (*High temp materials and HX design*)

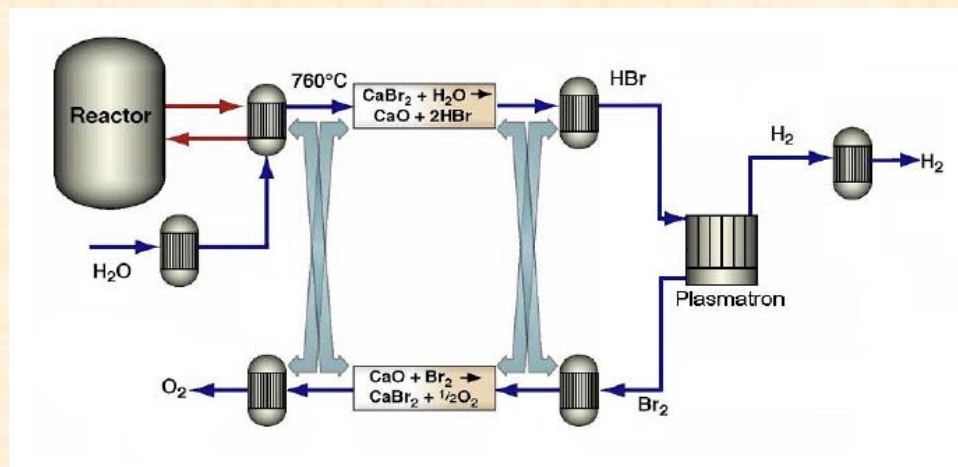


**S-I Thermochemical Cycle**

## Interface Technologies (HX, Materials)



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**Calcium Bromine Cycle**

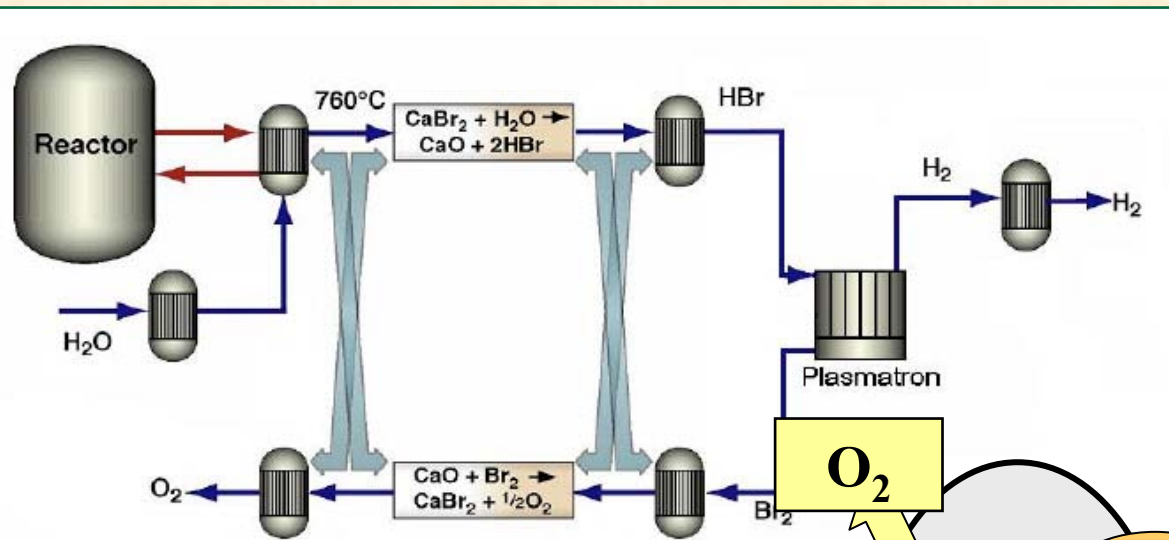
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# Other TC Cycles with Potential for Lower Temperature or Higher Efficiency Include:

- ❖ **Copper chlorine (GRI H-6)**
  - ◆ <550°C; low peak temperature
  - ◆ Issue: higher efficiency electrolysis
- ❖ **Iron chloride (Ispra Mark 9, GRI I-6)**
  - ◆ 650°C; low peak temperature
  - ◆ Issue: suppress competing chemical reaction
- ❖ **Iron chlorine (GRI B-1)**
  - ◆ 925°C; more mature
  - ◆ Issue: suppress competing chemical reaction
- ❖ **Copper sulfur (GRI H-5)**
  - ◆ 827-900°C; potential for higher efficiency
  - ◆ Issue: economics of scaling hybrid processes; higher efficiency electrolysis
- ❖ **Vanadium chlorine**
  - ◆ 925°C; full flowsheet available
  - ◆ Issue: high peak temperature; conflicting data on one reaction

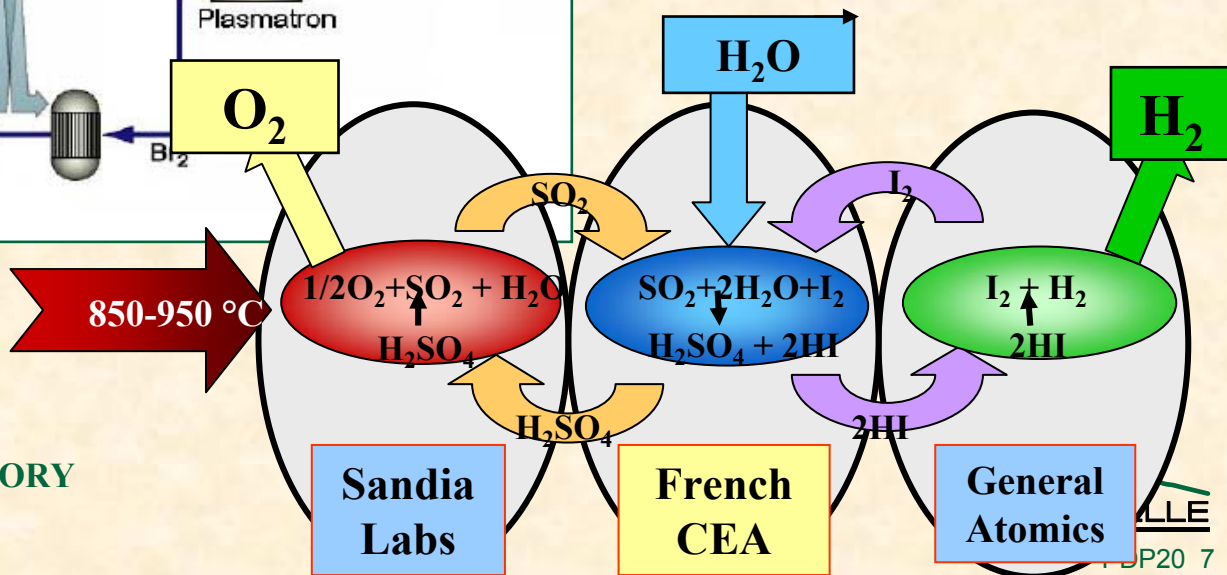
## Approach to Planning & Coordinating Includes:

- ❖ Understand the chemistry and temperature of various process steps
- ❖ Identify possible materials for use in these steps
- ❖ Establish test program to evaluate the materials/process



## Sulfur Iodine Cycle

- ❖ **HI section (GA)**
- ❖ **H2SO4 Section (SNL)**
- ❖ **Bunsen (CEA)**
- ❖ **Materials (INL, ORNL, Univ, Ind)**

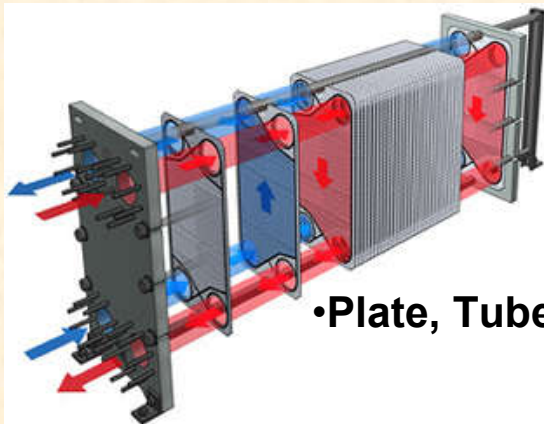


# **Materials for High-Temperature Thermochemical Processes - Technical Accomplishments Include:**

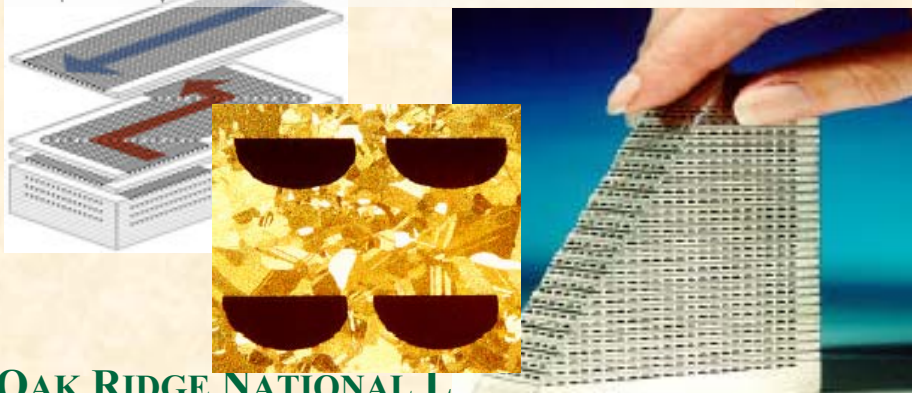
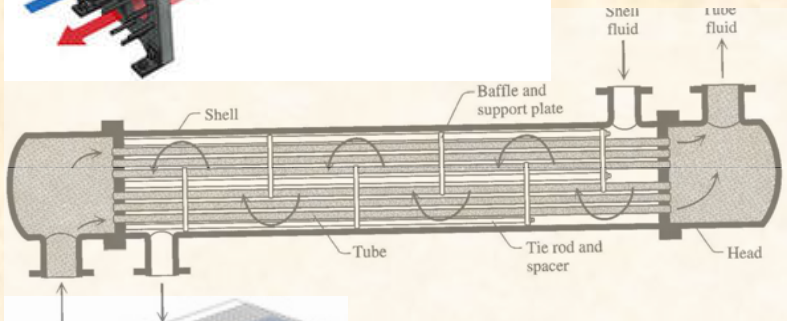
- ❖ **Completed assessment of NHI TC cycle service conditions, candidate materials**
- ❖ **Issued report: Materials Requirements for Nuclear Hydrogen Generation Systems**
  - ◆ **Identifies chemistry and temperatures for key process steps**
  - ◆ **Identifies candidate materials for major component in NHI TC cycles**
  - ◆ **Identifies testing approach and prioritizes materials R&D needs**



# Heat Exchanger Materials and Design Are Key Issues for Thermochemical Cycles

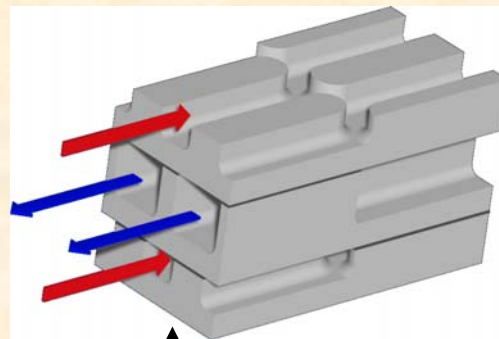


•Plate, Tube/Shell

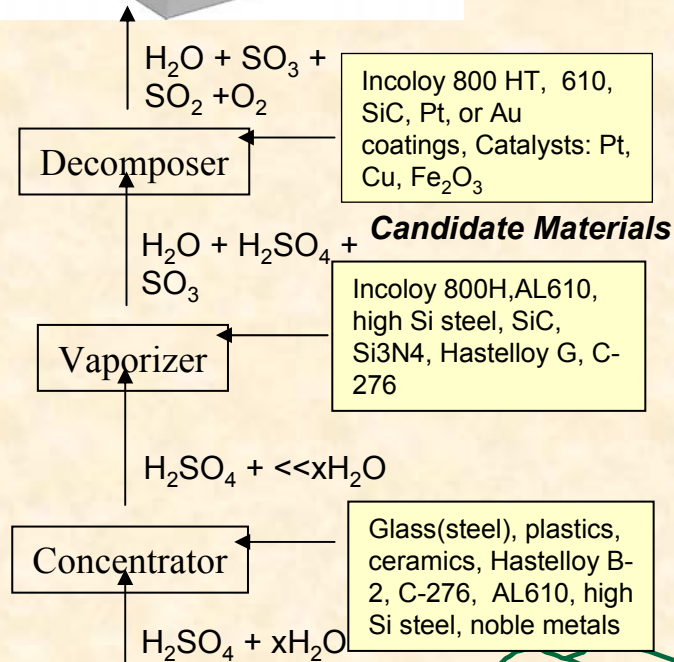


•Heatrix

## •Advanced Ceramic Materials (SiC, C-SiC)



Offset-fin  
C-SiC  
Composite  
plate heat  
exchanger



# Technical Accomplishment-Isolate Chemical and Temperature Environments ( $\text{H}_2\text{SO}_4$ & HI) Generation

## S-I Bunsen Reaction (120 – 130 °C; 0.1 to 0.3 MPa)

Stage	Environment (wt%)	Candidate material
Main Reaction	8 $\text{H}_2\text{O}$ ; 2 $\text{H}_2\text{SO}_4$ ; 80 $\text{I}_2$ ; 7 HI; 0.5 $\text{O}_2$ ; 1.7 $\text{SO}_2$	Ta, Zr, $\text{Si}_3\text{N}_4$ , $\text{SiO}_2$ , $\text{Al}_2\text{O}_3$ , B2, 242, Hastelloy C-276, and Nb-1Zr coating
$\text{H}_2\text{SO}_4$ Boost Reactor	57 $\text{H}_2\text{O}$ ; 43 $\text{H}_2\text{SO}_4$ ; 0.1 $\text{SO}_2$ ; trace $\text{I}_2$	
HI Phase $\text{SO}_2$ Stripper HX	trace $\text{H}_2\text{SO}_4$ ; 87 $\text{I}_2$ ; 7 HI; 0.2 $\text{SO}_2$ ; 6 $\text{H}_2\text{O}$ ;	
Vessel for Bunsen Reaction HXs	Ranges listed above	Fluorocarbon-lined (Teflon, Kynar, etc.) low alloy steels

# Technical Accomplishment-Isolate Chemical and Temperature Environments ( $\text{H}_2\text{SO}_4$ Concentration)

## S-I $\text{H}_2\text{SO}_4$ Concentrator

Stage	Temperature (°C)	Pressure (MPa)	Environment (wt%)	Candidate materials
Primary side	<450	0.1-6.8	He/molten salt	Hastelloy B2 & N, SiC, C-C composites, C276, 800/800H, Hi-Si steel (Duriclor 51M), glass-lined steel, Au, Pt, & Nb coatings
Secondary side	<400	0.1	0-0.1 $\text{SO}_2$ ; 57-98 $\text{H}_2\text{SO}_4$ ; 2-42 $\text{H}_2\text{O}$ ; trace $\text{I}_2$	

# Technical Accomplishment-Isolate

## Chemical and Temperature Environments ( $\text{H}_2\text{SO}_4$ Vaporizer)

### S-I $\text{H}_2\text{SO}_4$ Vaporizer

Stage	Temperature (°C)	Pressure (MPa)	Environment (wt%)	Candidate materials
Primary side	580 to 380	0.1-6.8	He/molten salt	Hastelloy B2, G & N, SiC, C-C composites, $\text{Si}_3\text{N}_4$ , C276, 800/800H, Hi-Si steel (Duriclor 51M), glass-lined steel, Au-, Pt-, & Nb-coatings
Secondary side	330 to 530	0.7	Liquid-Vapor 98-71 $\text{H}_2\text{SO}_4$ ; 0-22 $\text{SO}_3$ ; 2-7 $\text{H}_2\text{O}$ ;	



# Technical Accomplishment-Isolate Chemical and Temperature Environments ( $\text{H}_2\text{SO}_4$ Decomposer)

## S-I $\text{H}_2\text{SO}_4$ Decomposer

Stage	Temperature (°C)	Pressure (MPa)	Environment (wt%)	Candidate materials
Primary side	950 to 800	0.1-6.8	He/molten salt	Hastelloy B2, SiC, C-C composites, $\text{Si}_3\text{N}_4$ , 242, 214, C276, 800HT,
Secondary side	530 to 900	0.7	Inlet-Outlet 71-20 $\text{H}_2\text{SO}_4$ ; 22-13 $\text{SO}_3$ ; 7-16 $\text{H}_2\text{O}$ ; 0-40 $\text{SO}_2$ ; 0-10 $\text{O}_2$	Nb-1Zr, Au-, Pt-, Cu-, & glass-coatings, Pt-, Cu-, & $\text{Fe}_2\text{O}_3$ -catalysts

# Technical Accomplishment-Isolate Chemical and Temperature Environments (HI Decomposition)

## S-I Hlx Reactive Distillation

Stage	Temperature (°C)	Pressure (MPa)	Environment (wt%)	Candidate materials
Inlet feed stream	262	2.2	11 HI; 81 I <sub>2</sub> ; 8 H <sub>2</sub> O	Ta, Ta-10W, Mo, Nb, Nb-1Zr, Zircaloy 702, SiC, vitreous carbon, C-C composites, Bulk metallic glasses
Outlet column bottom	310	2.2	1 HI; 98 I <sub>2</sub> ; 1 H <sub>2</sub> O	
H <sub>2</sub> scrubber/ condenser	221 to 25	2.2	66 HI; 32 H <sub>2</sub> O; 2 H <sub>2</sub>	

# Technical Accomplishment-Isolate Chemical and Temperature Environments (HI Decomposition)

## S-I Hlx Phosphoric Acid Reactor

Stage	Temperature (°C)	Pressure (MPa)	Environment (wt%)	Candidate materials
Concentrated $\text{H}_3\text{PO}_4$	132 to 211	0.1	96 $\text{H}_3\text{PO}_4$ ; 4 $\text{H}_2\text{O}$	TBD based on relevant industrial experience
Custom column	120 to 241	0.3-0.9	74 $\text{H}_3\text{PO}_4$ ; 11 HI; 4 $\text{I}_2$ ; 10 $\text{H}_2\text{O}$	
Dilute $\text{H}_3\text{PO}_4$	250	0.95	87 $\text{H}_3\text{PO}_4$ ; 13 $\text{H}_2\text{O}$	
Iodine outlet	120	0.2-0.7	99.9 $\text{I}_2$ ; 0.1 $\text{H}_2\text{O}$	

# Materials Testing for NHI Thermochemical Cycles Has Been Initiated - HI Decomposition

Screening

Immersion  
Coupons

Development

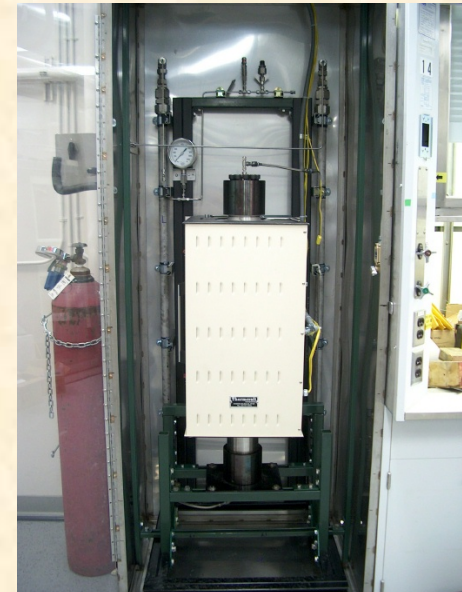
Crack Initiation & Growth,  
Long Term Testing,  
Cladding

Prototype

S-I Loop/Pilot Plant  
Testing

## ❖ *Work being performed by GA*

- ◆ S-I corrosion test facilities constructed for corrosion tests on selected coupons and stress specimens
- ◆ Screening tests underway for HI section materials
- ◆ Longer term testing, advanced materials activities being defined



Immersion corrosion test system II



# Materials Testing for $HI_x$ Section Has Begun

- ❖ Refractory metals and ceramics have shown the best corrosion performance to date

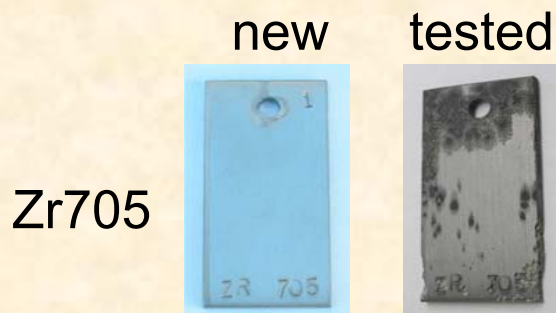
## ❖ $HI_x$ materials testing (UNLV and GA)

- ◆ 22 coupons from four classes of materials: refractory and reactive metals, superalloys and ceramics, have been screened.

Excellent	Good	Fair	Poor
Ta-40Nb, Nb-1Zr, Nb-10Hf, SiC(CVD), SiC(Ceram atec sintered), Mullite	Ta, Ta- 10W, Nb, Nb-7.5Ta, SiC (sintered) Si-SiC (3 kinds)	Mo-47Re, Alumina	Mo, C-276, Haynes 188, graphite*, Zr702, Zr705

\* structurally sound but absorbed  $HI_x$

- ◆ Long term corrosion performance testing has started
  - effect of  $HI_x$  on stress corrosion
  - cost reduction through cladding



# Technical Accomplishment-Isolate

## Chemical and Temperature

### Environments (Ca-Br Cycle)

Reaction Beds During HBr Production (0.1 MPa)

Stage	Temperature (°C)	Environment (wt%)	Candidate materials
HX primary side	750	He/molten salt	Hastelloy N, 232, 214, 713 cast, RA330
HX secondary side		100 CaBr <sub>2</sub> & H <sub>2</sub> O to 100 CaO & HBr	
Bed support			Ceramics/catalysts TBD
Reaction vessel	300 to 750	100 H <sub>2</sub> O & 0 HBr to 0 H <sub>2</sub> O & 100 HBr with trace O <sub>2</sub>	If insulated, Ni-Clad low alloy steel, If not, 713LC, 214, Ni <sub>3</sub> Al, MA956, MA754
Reaction vessel insulation	750		CaTiO <sub>3</sub> , Al <sub>2</sub> O <sub>3</sub>

# Technical Accomplishment-Isolate Chemical and Temperature Environments (Ca-Br Cycle)

## Reaction Beds During Regeneration (0.1 MPa)

Stage	Temperature (°C)	Environment (wt%)	Candidate materials
HX primary side	<590	He/molten salt	Hastelloy N, 232, 214, 713 cast, RA330
HX secondary side		100 Br <sub>2</sub> & CaO to 100 CaBr <sub>2</sub> & O <sub>2</sub>	
Bed support			Ceramics/catalysts TBD
Reaction vessel	200 to 590	100 Br <sub>2</sub> & 0 O <sub>2</sub> to 0 Br <sub>2</sub> & 100 O <sub>2</sub>	If insulated, Ni-Clad low alloy steel, If not, 713LC, 214, Ni <sub>3</sub> Al, MA956, MA754
Reaction vessel insulation	<590		CaTiO <sub>3</sub> , Al <sub>2</sub> O <sub>3</sub>

# Technical Accomplishment-Isolate Chemical and Temperature Environments (Ca-Br Cycle)

## Reaction Beds to Plasmatron HX (0.1 MPa)

Stage	Temperature (°C)	Environment (wt%)	Candidate materials
Primary side	750 to 50	100 HBr & 0.1 H <sub>2</sub> O to 3 Br <sub>2</sub> & 97 HBr with trace H <sub>2</sub> O	Ni, B2, 214, 232, Hastelloy N, stainless steel, Si <sub>3</sub> N <sub>4</sub> , Nb-1Zr
Secondary side	25	H <sub>2</sub> O	

## Plasmatron HX (0.1 MPa)

Stage	Temperature (°C)	Environment (wt%)	Candidate materials
Inlet	50	3 Br <sub>2</sub> & 97 HBr with trace H <sub>2</sub> O	300 series stainless steel
Outlet	<<300	50 HBr, 25 Br <sub>2</sub> , & 25 H <sub>2</sub>	



# Technical Accomplishment-Isolate Chemical and Temperature Environments (Ca-Br Cycle)

## Other Components

Stage	Temperature (°C)	Pressure (MPa)	Environment (wt%)	Candidate materials
Multistage compressors	100 to 335	0.005 to 3.5	30 H <sub>2</sub> , 4 Br <sub>2</sub> , & 2 H <sub>2</sub> O	TBD based on industrial experience
Steam superheater Primary side	750 to 850	6.8-0.1	He/molten salt	617, 625, 230, B2, 214, 242, Hastelloy N
Steam superheater Secondary side	750	0.1	H <sub>2</sub> O	

# Summary of High Priority R&D

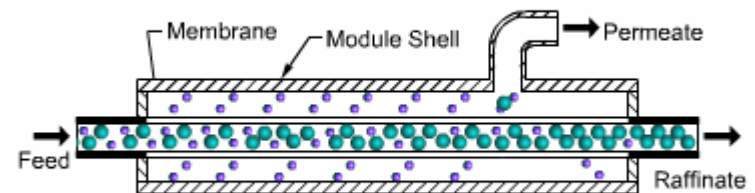
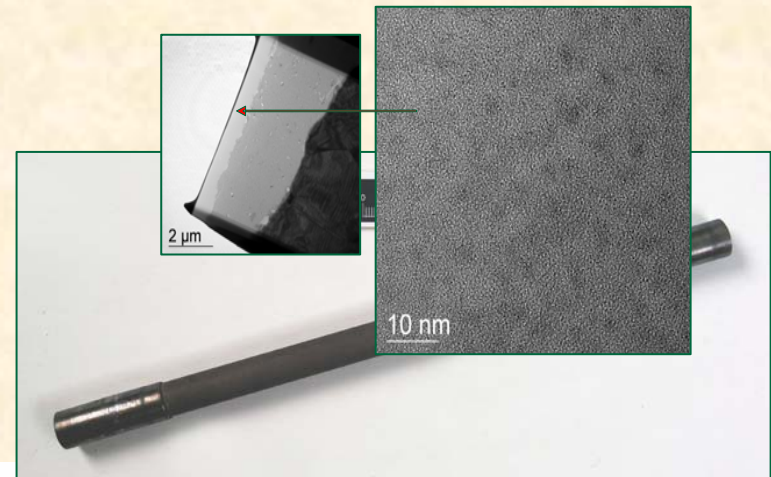
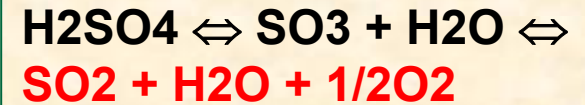
❖ Prioritized high due to technical viability issues associated with high temperature/corrosive environments

◆ Sulfur Cycles (S-I, Hybrid S)

- $\text{H}_2\text{SO}_4$  concentrator, vaporizer
- HI reactive distillation column
- Inorganic membranes
  - Potential for increased product formation
  - Potential for reduced operating temperature

◆ CaBr

- Reactive bed HX
- Ca-Br/HBr HX
- Reaction bed vessel



# Future Work

- ❖ **FY05 – Revise and update materials selection document**
- ❖ **FY06 - Develop a prioritized, integrated materials evaluation program**
  - ◆ **So as to establish engineering feasibility**

# Publications and Presentations

- ❖ **Materials Requirements for Nuclear Hydrogen Generation Systems, ORNL TM, 2004**
- ❖ **W. R. Corwin, *NHI Materials Selection and R&D Plan*, NHI Materials Planning Meeting, UNLV, Las Vegas, Nevada, May 17, 2004**
- ❖ **W. R. Corwin, *NHI Materials Selection and R&D Plan & NHI Membrane Studies Update*, NHI Semiannual Program Review, DOE HQ, Germantown, Maryland, September 21-22, 2004**
- ❖ **W. R. Corwin, *High Priority Materials R&D for NHI*, NHI Materials Program Review Meeting, ORNL, February 10, 2005**



# Hydrogen Safety

**What is the most significant hydrogen hazard associated with this project?**

- ❖ Hydrogen explosiveness/flammability (4 to 75 vol.%) during performance evaluation

**What are you doing to deal with this hazard?**

- ❖ Work will be performed under the safety envelopes of the various testing organizations
- ❖ Use both administrative and physical controls
  - ◆ Work review/authorization
  - ◆ Volume/pressure controls
  - ◆ Sensors